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## **Citation:**

2-Furyl Phosphines as Ligands for Transition-Metal-Mediated Organic Synthesis Neil G. Andersen and Brian A. Keay pp 997 – 1030.

## **Charts:**

Chart 1. The Tolman Cone Angle

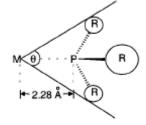


Chart 2. Equations Describing Complexation of TFP with  $Pd(dba)_2$ 

 $Pd(dba)_2 + 2TFP \longrightarrow Pd(dba)(TFP)_2 + dba$  (1)

Pd(dba)(TFP)2 + S - SPd(TFP)2 + dba

$$\frac{K_1 = [SPd(TFP)_2][dba]}{[Pd(dba)(TFP)_2]}$$
(2)

$$Pd(dba)(TFP)_{2} + TFP \xrightarrow{\bullet} SPd(TFP)_{3} + dba$$

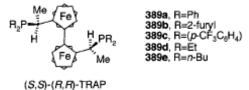
$$K_{0} = \underline{[SPd(TFP)_{3}][dba]} \\ \underline{[Pd(dba)(TFP)_{2}][TFP]}$$
(3)

$$SPd(TFP)_{3} \xrightarrow{\bullet} SPd(TFP)_{2} + TFP$$

$$K_{2} = \underline{[SPd(TFP)_{2}][TFP]} = K_{1}/K_{0} \qquad (4)$$

$$SPd(TFP)_{2} + ArX \xrightarrow{\bullet} ArPdX(TFP)_{2} + S \quad k \qquad (5)$$

Chart 3. Selected Trans-Chelating Chiral Diphosphine Ligands (TRAPs)



# **Tables:**

ligand	cone angle ( $\theta^{c}$ )	ligand	cone angle ( $\theta$ )
PH <sub>3</sub>	87°	P(p-Tol) <sub>3</sub>	145°
P(OMe)3	107°	P(m-Tol) <sub>3</sub>	165°
PMe <sub>3</sub>	118°	PCy <sub>3</sub>	170°
P(OPh) <sub>3</sub>	128°	P(O-t-Bu)3	172°
PEt <sub>3</sub>	132°	$P(t-Bu)_3$	182°
TFP	133°	$P(C_6F_5)_3$	184°
$P(CF_3)_3$	137°	P(o-Tol) <sub>3</sub>	194°
PPh <sub>3</sub>	145°	P(mesityl) <sub>3</sub>	212°

Table 1. Tolman Cone Angle  $(\theta')$  for a Variety of Phosphines and Phosphinites^3

Table 2. Electronic Parameter v for a Variety of Phosphines and Phosphinites<sup>8</sup>

ligand	$\nu$ (cm <sup>-1</sup> )	ligand	ν (cm <sup>-1</sup> )
PF3	2110.8	P(p-Tol) <sub>3</sub>	2066.7
P(C <sub>6</sub> F <sub>5</sub> ) <sub>3</sub>	2090.9	P(o-Tol) <sub>3</sub>	2066.6
P(OPh) <sub>3</sub>	2085.3	PMe <sub>3</sub>	2064.1
PH <sub>3</sub>	2083.2	PEt <sub>3</sub>	2061.7
P(OMe) <sub>3</sub>	2079.5	PBu <sub>3</sub>	2060.3
PPh <sub>3</sub>	2068.9	PCy <sub>3</sub>	2056.4
P(m-Tol) <sub>3</sub>	2067.2	P(t-Bu) <sub>3</sub>	2056.1

Table 3. <sup>31</sup>P-<sup>77</sup>Se Coupling Constants for Various Phosphine Selenides (R<sub>3</sub>P=Se)<sup>9</sup>

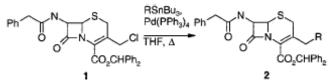
$PR_3$	$^{1}J$ (Hz)	PR <sub>3</sub>	1J (Hz)
P(p-MeOC <sub>6</sub> H <sub>4</sub> ) <sub>3</sub>	708	PPh <sub>2</sub> (2-furyl)	754
PPh <sub>2</sub> (o-Tol)	730	P(2-thienyl) <sub>3</sub>	757
PPh <sub>3</sub>	732	$PPh_2(m-CF_3C_6H_4)$	766
PPh <sub>2</sub> (2-thienyl)	743	PPh(2-furyl) <sub>2</sub>	774
PPh(2-thienyl)2	752	P(2-furyl) <sub>3</sub>	793

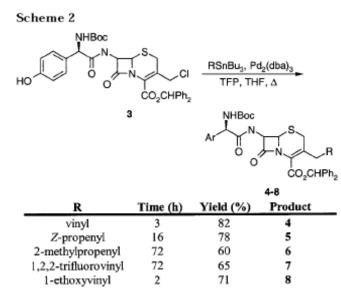
Table 4. Relative Rates of Stille Coupling between Iodobenzene and Vinyltributyltin with Various  $Pd_2(dba)_3/Ligand$  Catalysts at 50 °C in THF<sup>10</sup>

	ligand <sup>a</sup>	cone angle	relative rate <sup>b</sup>	inhibition factor <sup>e</sup>	yield (%) <sup>d</sup>
1	PPh <sub>3</sub>	145°	1	19	15.2
2	(p-Tol) <sub>3</sub> P	145°	< 0.07	>100	<2
3	(o-Tol) <sub>3</sub> P	194°	35.2	3.4	19
4	TFP	133°	105	3.7	>95
5	P(C6F5)3	184°	e		13.2
6	Ph <sub>2</sub> As	142°	1100	1.3	>95

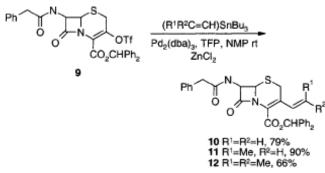
 $^{a}$  Pd:L ratio = 1:4.  $^{b}$  For PPh<sub>3</sub>,  $k = 4.6 \times 10^{-5}$  min<sup>-1</sup>.  $^{c}$  Ratio of PdL<sub>2</sub> catalyst rate to PdL<sub>4</sub> catalyst rate.  $^{d}$  HPLC yield after 72 h,  $^{o}$  Catalyst decomposition was instantaneous (<2 min).

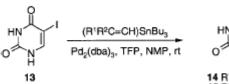
## **Schemes:**





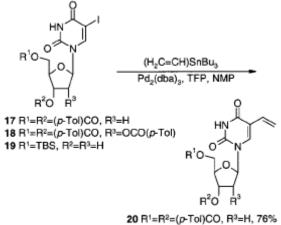






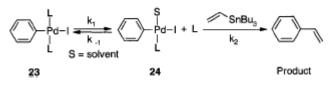


14 R<sup>1</sup>=R<sup>2</sup>=H, 89% 15 R<sup>1</sup>=Ph, R<sup>2</sup>=H, 92% 16 R<sup>1</sup>=H, R<sup>2</sup>=Me, 70%

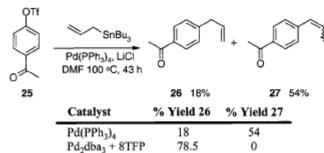


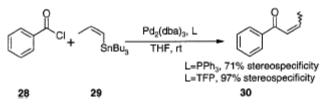
21 R<sup>1</sup>=R<sup>2</sup>=(*p*-Tol)CO, R<sup>3</sup>=OCO(*p*-Tol), 98% 22 R<sup>1</sup>=TBS, R<sup>2</sup>=R<sup>3</sup>=H, 91%

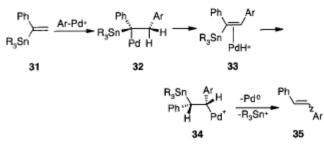
Scheme 6



Scheme 7



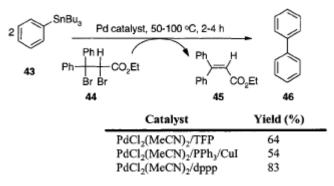


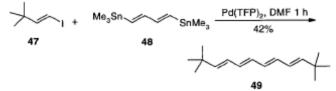


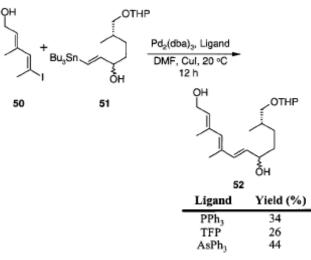
Scheme 10			
Me SnBu <sub>3</sub>	+	Pd <sub>2</sub> (dba) <sub>3</sub> , Li NMP, cat. Ci	
36	37	,сн₂он	Ar, ,CH₂OH
	Me	e Ar *	Me
	38 5	Stille Product	39 cine Product

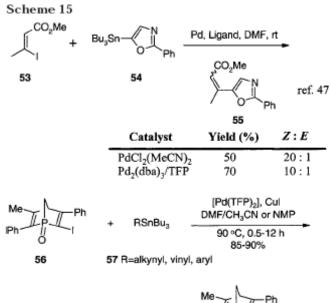
Ligand	Time (h)	Conversion (%)	Stille:cine
TFP	90	30	4:1
Ph <sub>3</sub> As	160	>90	2:1

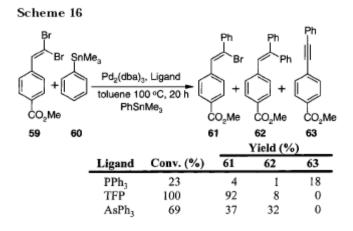
	CoDu	f, Pd <sub>2</sub> (dba) <sub>3</sub> , L LiCl, rt, 48 h	
40	TMS	Ph + TMS	Ph
		41	42
Ligand (L)	Yield (%)	41 (% E isomer)	α:γ
Ligand (L) PPh <sub>3</sub>	Yield (%) 66	41 (% E isomer) 15	α:γ 61:39

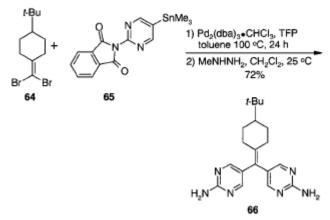


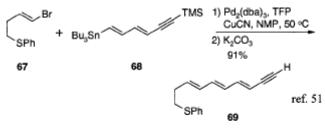


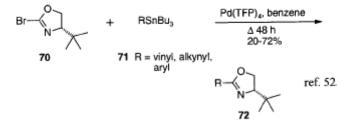


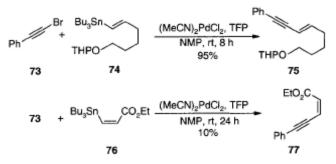


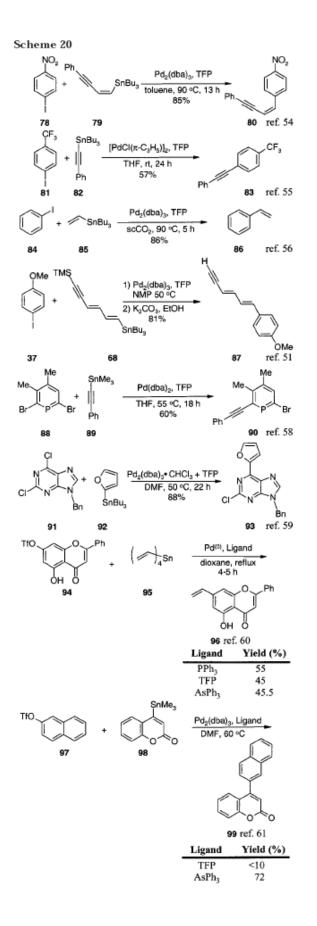


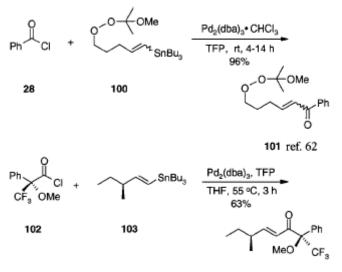






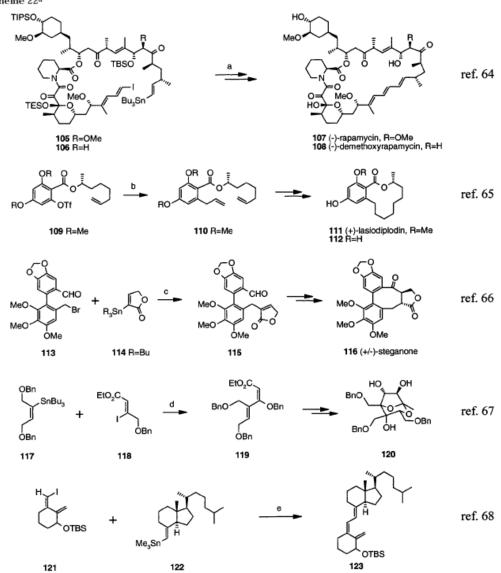






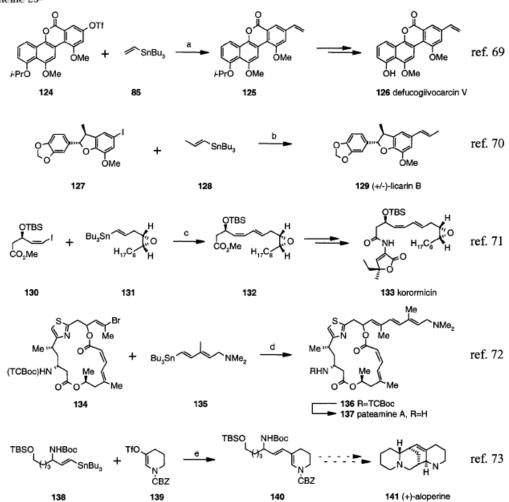
104 ref. 63



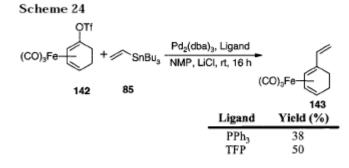


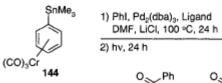
<sup>a</sup> Conditions: (a) (TFP)<sub>2</sub>PdCl<sub>2</sub>, *N*,*N*-diisopropylethylamine (DIPEA), DMF, THF, rt, 74% R = OMe; 65% R = H. (b) allyltributylstannane, LiCl, Pd<sub>2</sub>(dba)<sub>3</sub>, TFP, 1-methyl-2-pyrrolidinone (NMP), 40 °C, 93%. (c) Pd<sub>2</sub>(dba)<sub>3</sub>, TFP, *N*,*N*-dimethylacetamide (DMA), 80 °C, 83%. (d) Pd<sub>2</sub>(dba)<sub>3</sub>, TFP, ZnCl<sub>2</sub>, DMF, 50 °C, 4.5 days, 86%. (e) Pd<sub>2</sub>(dba)<sub>3</sub>, TFP, CuI, DMF, 25 °C, 4 days, 33%.

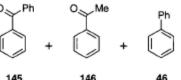
Scheme 23<sup>a</sup>



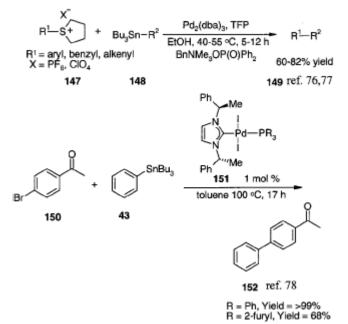
<sup>a</sup> Conditions: (a) Pd<sub>2</sub>(dba)<sub>3</sub>, TFP, NMP, rt, 5 h, 69%. (b) Pd<sub>2</sub>(dba)<sub>3</sub>, TFP, LiCl, DMF, 120-130 °C, 84-86%. (c) Pd<sub>2</sub>(dba)<sub>3</sub>·CHCl<sub>3</sub>, TFP, NMP, rt, 6 days, 34%. (d) Pd<sub>2</sub>(dba)<sub>3</sub>, TFP, NMP, 25 °C, 27% (57% based on recovered starting material); TCBoc = 1,1-dimethyl-2,2,2-trichloroethoxycarbonyl. (e) Pd<sub>2</sub>(dba)<sub>3</sub>, TFP, LiCl, NMP, rt, 1.5 days, 93%.

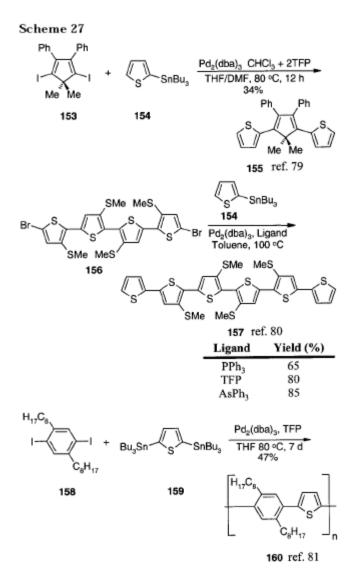


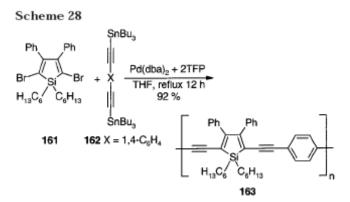


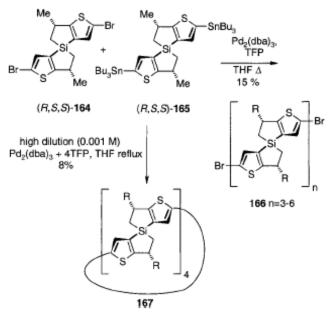


145	140	Yield (%	40	
Ligand	145	146	46	
PPh <sub>3</sub>	44	24	13	
TFP	36	25	2	

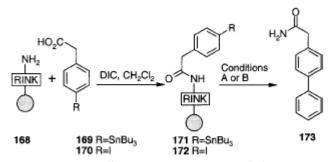






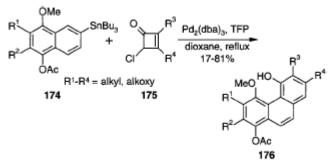


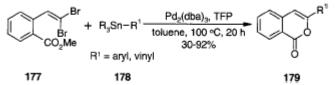
Scheme 30<sup>a</sup>

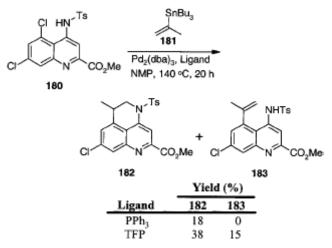


<sup>a</sup> Conditions A: (i) 3 equiv of iodobenzene (84), 10 mol % Pd<sub>2</sub>(dba)<sub>3</sub>, 10 mol % TFP, 2 equiv of LiCl, NMP, 25 °C, 12 h; (ii) 5% TFA-CH<sub>2</sub>Cl<sub>2</sub>, 15% yield (2 steps). Conditions B: (i) 3 equiv of trimethylphenyltin, 10 mol % Pd<sub>2</sub>(dba)<sub>3</sub>, 10 mol % TFP, 2 equiv of LiCl, NMP, 25 °C, 12 h; (ii) 5% TFA-CH<sub>2</sub>Cl<sub>2</sub>, 33% yield (2 steps).

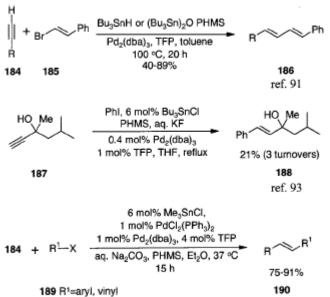


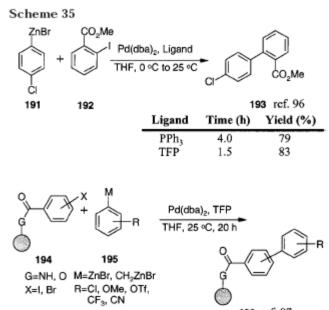




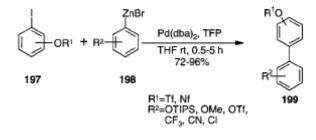


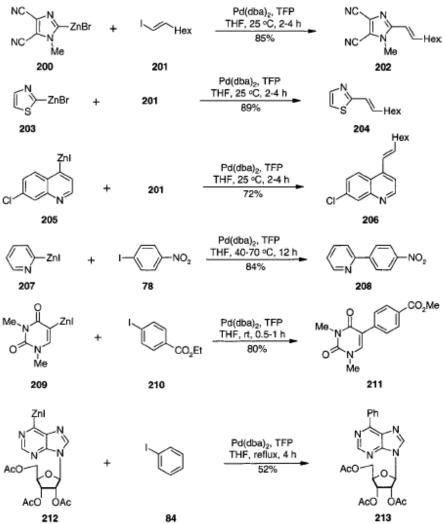


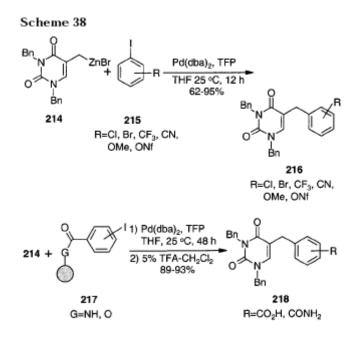








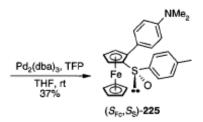


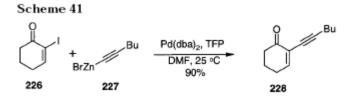


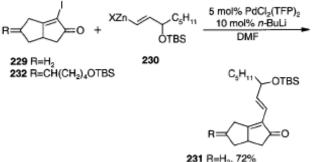






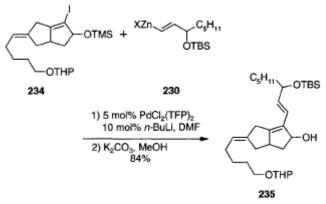


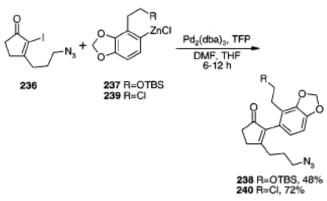


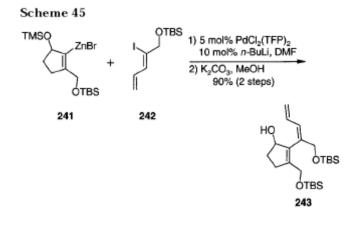


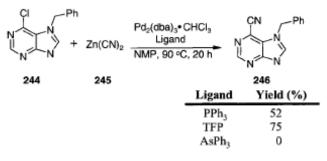
231 R=H<sub>2</sub>, 72% 233 R=CH(CH<sub>2</sub>)<sub>4</sub>OTBS, <5%

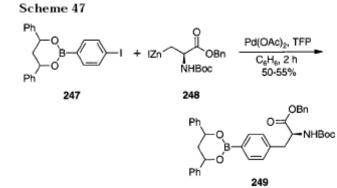


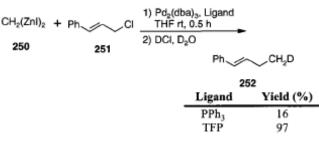




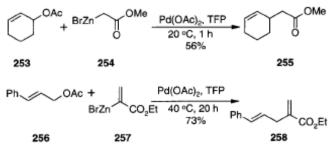


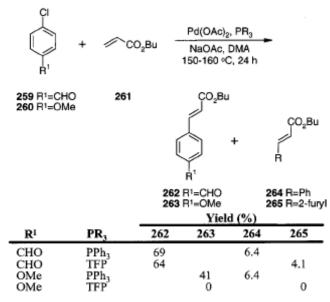


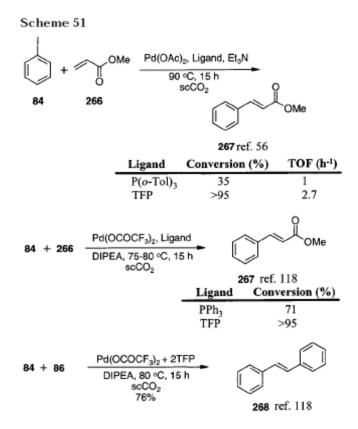


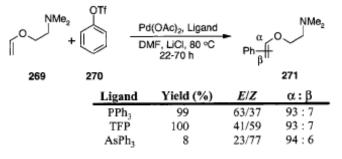




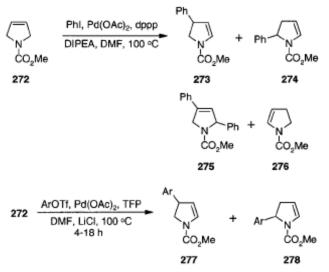




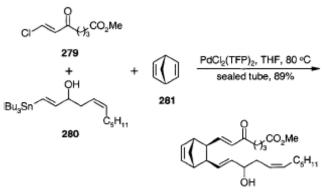


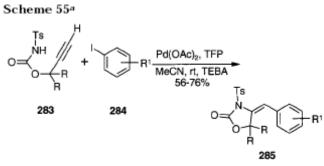




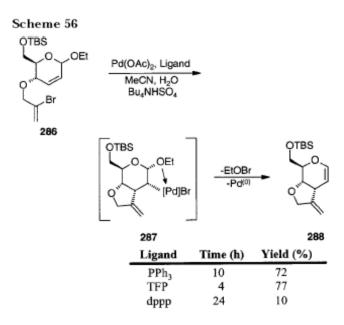


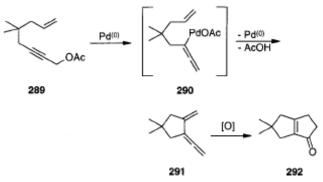


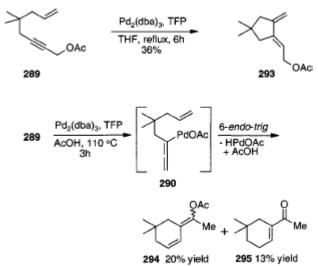


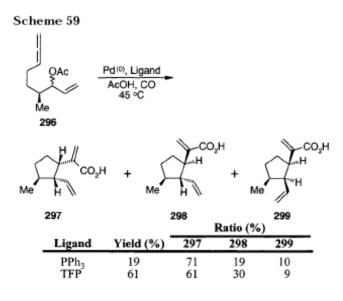


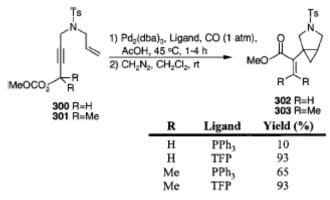




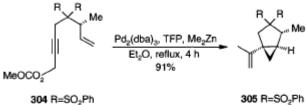








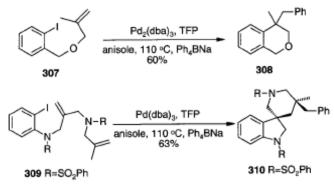
Scheme 61



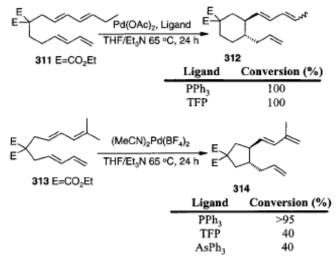
305 R=SO<sub>2</sub>Ph

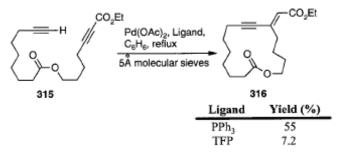
Лe

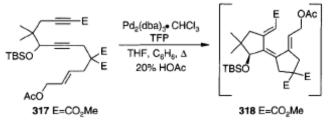
306 (-)-α-thujone

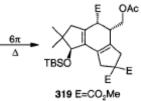




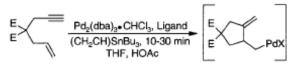






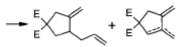


Scheme 66



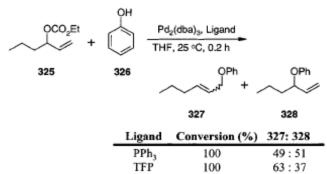
320 E=CO2Bn

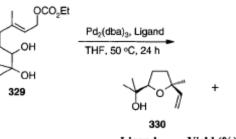
321 E=CO<sub>2</sub>Bn

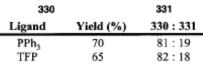


322 E=CO <sub>2</sub> Br	323/324 E=CO2Bn	
	Yield (%)	
Ligand	322	323/324
PPh <sub>3</sub>	28	30
TFP	trace	69
none	45	0

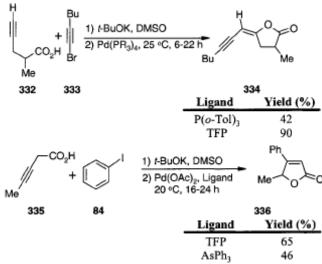


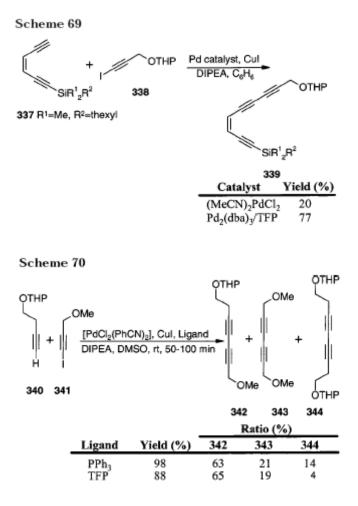


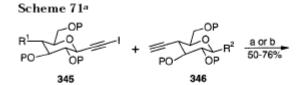








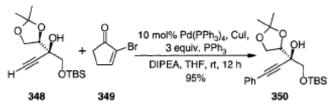


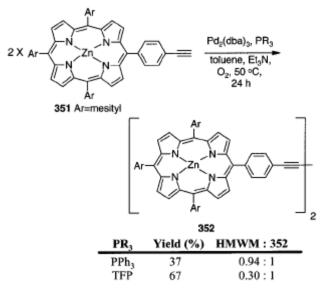




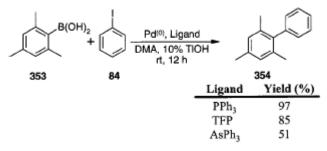
<sup>a</sup> Conditions: (a) PdCl<sub>2</sub>(PhCN)<sub>2</sub>, TFP, DIPEA, DMSO, 50 °C. (b) Pd<sub>2</sub>(dba)<sub>3</sub>, TFP, CuI, DMSO, rt.



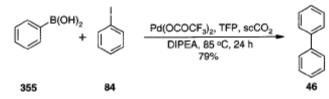


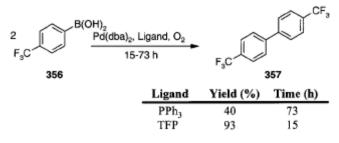


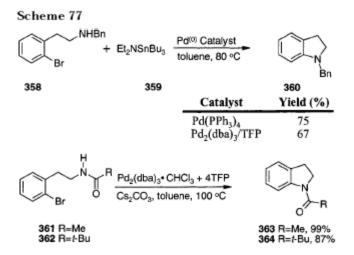
Scheme 74



Scheme 75

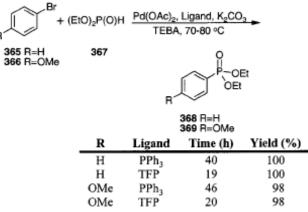




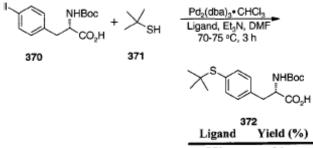


Scheme 78<sup>a</sup>

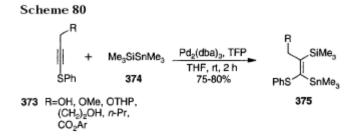
B

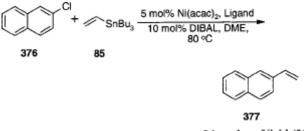


<sup>a</sup> TEBA = benzyltriethylammonium chloride.



Ligand	Yield (%)
PPh <sub>3</sub>	26
TFP	trace
dppf	99

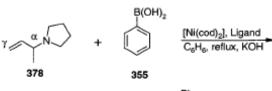




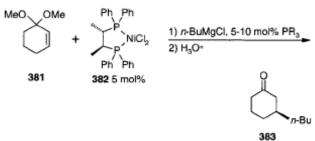
•

Ligand	Yield (%)
PPh <sub>3</sub>	86
TFP	64

Scheme 82

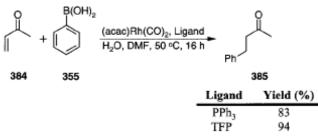


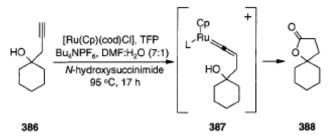
55	h + Ph	~~r
379		380
Ligand	Yield (%)	α:γ
PPh <sub>3</sub>	72	1.6:1
TFP	54	1.6:1



PR3	Conversion (%)	ee (%)
PPh <sub>3</sub>	61	82
TFP	60	40



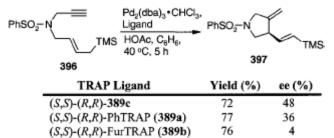




#### Scheme 86

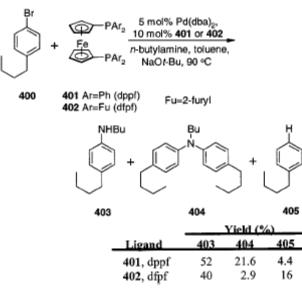
N H2 (1 kg/cm/	6, TRAP Ligano ), 50 °C, 24 h, I <sub>2</sub> CH <sub>2</sub> CI	•	
390 R=i-Bu			391
TRAP Ligand	Conv. (%)	ee (%)	Config.
(S,S)-(R,R)-PhTRAP (389a)	100	90	S
(S,S)-(R,R)-FurTRAP (389b)	100	35	S

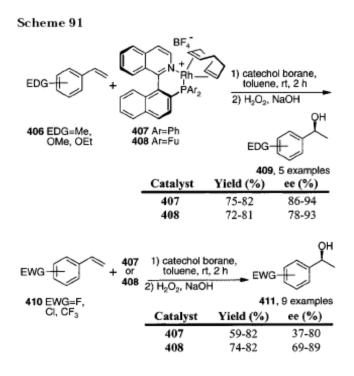
MeO.C. TRAP	d) <sub>2</sub> ]BF <sub>4</sub> , Ligand m <sup>2</sup> ), CH <sub>2</sub> Cl <sub>2</sub> , 3-6 h	► MeO <sub>2</sub> C	CO <sub>2</sub> Me
TRAP Ligand	Conv. (%)	ee (%)	Config.
(R,R)-(S,S)-EtTRAP (389d)	100	96	S
(R,R)-(S,S)-PhTRAP (389a)	100	26	R
(R,R)-(S,S)-FurTRAP (389b)	100	7	R
$MeO_2C \underbrace{\bigcirc}_{CO_2Me} \underbrace{\xrightarrow{[Rh(cod)_2]BF_4,}_{TRAP Ligand}}_{IPA, 20 \circ C, 24 h} MeO_2C \underbrace{\bigcirc}_{CO_2Me} \underbrace{\bigcirc}_{CO_2Me}_{395}$			
TRAP Ligand	Conv. (%)	ee (%)	Config.
(R,R)-(S,S)-BuTRAP (389d)	98	71	S
(R,R)-(S,S)-PhTRAP (389a)	trace		
(R,R)-(S,S)-FurTRAP (389b)	13	6	R

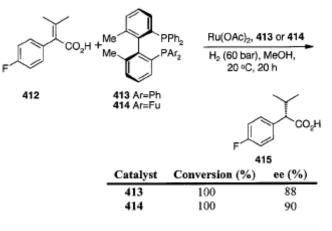


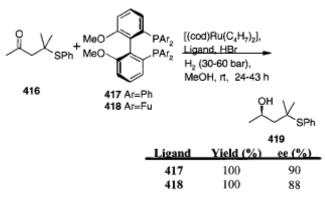
#### Scheme 89

Î /	O 1) [Rh(cod) <sub>2</sub> ]BF <sub>4</sub> , TRAP Ligand ↓ Ph <sub>2</sub> SiH <sub>2</sub>		
398	2) K <sub>2</sub> CO <sub>3</sub> , MeOH		он 399
TRAP Ligand		dl:meso	ee (%)
(R,R)-(S,S)-FurTRAP (389b)		86:14	91
(R,R)-(S,S)-PhTRAP (389a)		49:51	29

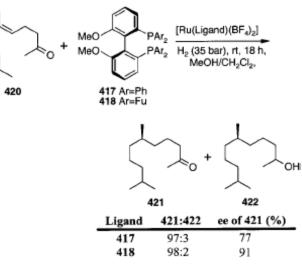


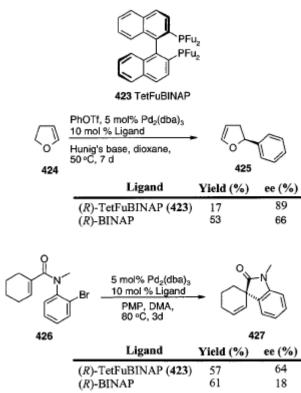


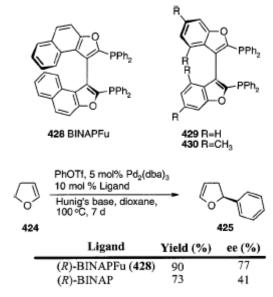












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